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Research Article

# Concentration of heavy metals in two fish species (*Cynoscion regalis* and *Pomatomus saltatrix*) from an oil drilling area in Western Coast of Ghana and public health risk assessment

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#### Abstract

This study assessed heavy metals (Lead (Pb), Nickel (Ni), Iron (Fe), Chromium (Cr), and Cadmium (Cd)) concentration in water (from borehole, hand-dug well and the sea) and fish (*Cynoscion regalis* and *Pomatomus saltatrix* species) and possible health risk exposure by the consumption of these fishes. A total of 18 samples of water and samples of two species of fish (*C. regalis* and *P. saltatrix*) were collected from Jomoro, Ellembelle and Ahanta West District and analyzed for the concentration of Pb, Ni, Fe, Cr, and Cd using Atomic Absorption Spectrometer (AAS). Results revealed appreciable concentrations of Pb, Ni, and Fe in the fish muscle and Ni, Cr, and Fe in water. Cadmium and Cr were not detected in the fish samples. Heavy metals in fish increased in the order Pb < Ni < Fe. The Estimated Daily Intake of fish value for adult ranged 1.88 x 10<sup>-3</sup> to 2.03 x 10<sup>-1</sup> (mg/kg bw/day), however, those for children were 4.40 x 10<sup>-3</sup> to 4.74 x 10<sup>-1</sup>(mg/kg bw/day). For health risk assessment, the Total Targeted Hazard Quotient for adults was < 1; however, those for children were >1 for *C. regalis* but less than 1 (< 1) for *P. saltatrix* meaning children who consume *C. regalis* are exposed to health risk in their life time. The results confirmed that contaminated environments could result in bioaccumulation of metal pollutants in fish's muscle, which poses a health risk to humans when consumed. Awareness and education of the general public on the threshold of heavy metals in fish and water are crucial to safeguard against health risks.

Keywords: Cynoscion regalis, Heavy metals, Health risk assessment, Oil, Pomatomus saltatrix, Water

# INTRODUCTION

Heavy metals remain one of the important classes of environmental pollutants due to their toxicity and persistence in the environment. Though heavy metals occur as natural elements, studies relate their increasing concentrations and pollution to point source anthropogenic and industrial activities, including oil and gas extraction (Noor-ul and Tauseef 2015; Faz and Acosta 2017; Demková *et al.*, 2017 and Kinuthia *et al.*, 2020). Analy-

sis of crude oil shows relatively high concentrations of heavy metals such as Cu, Hg, Zn, Pb and Fe (Kadafa, 2012). During oil and gas production, hydrocarbon reservoirs release a by-product (Produced Water) which is either disposed-off into the ocean or may be re-injected into the wells to promote oil recovery (Igwe *et al.*, 2013). The inability to eliminate heavy metals through natural processes aggravates the situation. This results in heavy metals moving from one compartment of the environment (aquatic) to another (terrestrial and biota)

(Katavut et al., 2010) with detrimental impacts like bioaccumulation. Exposure to high concentrations of Pb, Ni and Cr are reported to cause central nervous system disorders, cardiovascular diseases, kidney failure, lung fibrosis, fertility problems, sinus node dysfunction and atrioventricular conduction disturbances, liver necrosis, myocardial dysfunction and cardiogenic shock among others.

In adjoining drilling communities, heavy metals contamination in water bodies might result from the indiscriminate discharge of drilling effluents and accidental leakages through iron pipes, thus contributing directly to heavy metals concentrations in the aquatic environment (Owamah 2013 and Ghaderpoori et al., 2018). Pollution of such nature has direct and indirect devastating effects on biological, physical and chemical characteristics of the environment (Abdel-Shafy and Mansour, 2016) with biodiversity and ecosystem services loss and, threats to public health (Ubiogoro and Adeyemo, 2017). It is established for example that, the presence of oil in water results in oxygen deficit thus affecting the rate of fish metabolism and decreases their ability to withstand organic and inorganic poisons (Lloyd, 1992). Other impacts include disruptions in response patterns associated with stimulation, depression, movement, coordination, and breathing, leading to the death of organisms (Demeke and Tassew 2016).

In Ghana, the oil and gas industry became prominent in 1970 along the Western Coast off the Cape Three Points although exploratory efforts began in 1896 (GNPC, 2009). A rapid development in 2010 paved the way for pumping to start from wells in the Deepwater Tano and the West Cape Three Points blocks making Ghana an important producer in West Africa after Nigeria, "pumping upwards of 500,000 barrels per day" (Global Muckraking, 2012). The heightened oil production and related activities in the deep waters of Ghana have raised concerns regarding the potential pollution of the aquatic system and its implications for the communities in the activity zone who are largely involved in fisheries and agriculture-related livelihoods.

To effectively track and manage the impact of oil and gas exploration on the communities along the coast, it is crucial to understand the dynamics and spatial distribution of pollutants such as heavy metals concentration in the environment. However, there is little information on the concentration of heavy metals in water and fish in communities along the oil and gas drilling areas of Cape Three Points in Western Ghana. Preliminary analysis of polycyclic aromatic hydrocarbons (PAHs) concentrations in water showed results below the detection limit. This research aims at assessing the concentrations of heavy metals in water and fish and the public health risks associated with the consumption of fish from the western coast of Ghana.

## **MATERIALS AND METHODS**

### Study area

The study was conducted in three districts along the western coast of Ghana near the jubilee oil field in the Western Region of Ghana. Six communities were randomly selected along the coastline within 60km from where oil and gas activities occur. The chosen communities were Half Assini and Kabenlasuazo in Jomoro District, Atuabo and Eikwe in Ellembelle District and Cape Three Points and Princess Town in Ahanta West District (Fig. 1). In each community, two sites were identified purposively for sampling. The selection criterion was based on distance (60km) to the area of oil and gas exploration. The study area is located between latitude 4°.45'N - 5°.02'N and longitude 2°.04'W - 2°.53'W and is in close proximity to Ghana's major offshore oil and gas exploration and production sites. The sandy shores of the Cape Three Points area serves as nesting grounds for sea turtles and habitats for several species, including the ghost crab, isopods, polychaetes, bivalves and gastropods (Armah et al., 2004).

### Water sampling and physico-chemical analysis

Water samples were collected from 6 hand-dug wells, 6 boreholes (drinking water sources) and sea water (six samples) located within the study area. Wells and boreholes were selected based on the distance to gas and oil drilling area. The control samples were, however, collected from areas where no oil and gas activities take place (about 13.6km away from the experimental site). Two water samples were collected from each category of water sources from each community within the study area. Similarly, two samples were collected from the control. A water sampler was used to collect water samples into 1.5 L and 500 mL pre-cleaned polyethylene sample bottles with caps for heavy metals and physico-chemical analysis, respectively. The sampling bottles were rinsed with the water source before taking the water samples. The samples were well labelled and transported to the Ecological Laboratory (ECOLAB) of University of Ghana, Legon, on ice in clean ice chests and stored in the refrigerator at 4 °C until they were analysed for physico-chemical parameters and heavy metals. The physico-chemical parameters analysed included: pH, electrical conductivity, biological oxygen demand (BOD), dissolved oxygen (DO), dissolved solids (TDS), turbidity and salinity. All the analytical procedures used in the physico-chemical analysis of the water samples were done according to the standard method of water analysis (Rodier et al., 2009).

### Fish samples collection

Samples of the most common fish species (*Pomatomus saltatrix* (Blue fish) and *Cynoscion regalis* 

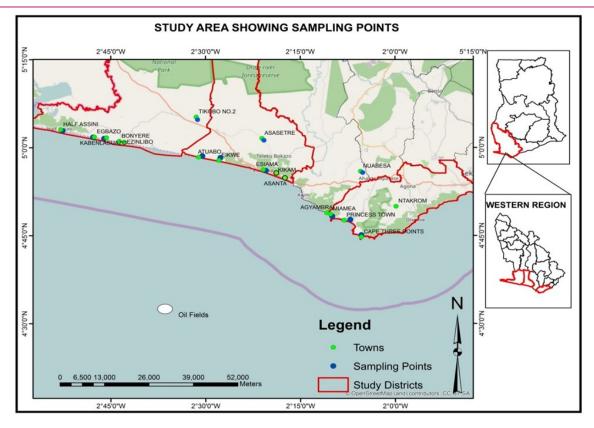


Fig. 1. Map of the study area.

(Weakfish) were obtained from local commercial fishermen in March 2017. Samples for the control were collected outside the study area. The fish samples were well labelled and kept in an ice chest at 4°C to avoid microbial activity and were transported to the Ecological Laboratory of the University of Ghana for analysis of heavy metals.

# Determination of heavy metals in water and fish samples

Standard procedures were followed for the quantitative determination of heavy metals in all samples. Heavy metal concentrations in the water and fish (Iron (Fe), Lead (Pb), Cadmium (Cd), Nickel (Ni) and Chromium (Cr)) samples were determined using direct flame Atomic Absorption Spectrometry (AAS) of model PinAAcle 900T, Boston, MA, USA by using different Cathode lamps.

Concentrations of heavy metals in water samples were determined using a specific wavelength for each metal. Average values of three replicates were taken for each determination. An appropriate drift blank was taken before the analysis of samples. The working wavelength for the heavy metals was 248.3 nm for Fe, 217 nm for Pb, 228.8 nm for Cd, 357.9nm for Cr and 232 nm for Ni.

Each of the fish samples was dissected using stainless steel scalpels, forceps and scissors in a dissecting tray to separate muscles and gills. The muscles and gills were then dried in labelled pre-washed crucibles at 70 °C for 24 h in an oven. Each dried sample was then pulverised, homogenized with a porcelain mortar and pestle and stored in labelled pre-treated specimen bottles, then transferred to preweighed crucibles and dried again oven until a constant weight was obtained. One gram of the dry sample was placed in a Teflon beaker and digested with 10 mL analytical grade nitric acid (HNO<sub>3</sub>) and hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) (1:1) on a hot plate at 100 °C to near dryness when a clear solution was obtained. The solution was then diluted up to 25 mL with distilled water, following which the concentrations of trace elements in each sample were determined using atomic absorption spectrophotometer. Blank experiments were run to check for background contaminants by the reagents and apparatus used. The values obtained from running blank experiments were subtracted from the analyte values as applicable. The Atomic Absorption Spectrophotometer (AAS) was precalibrated. The calibration was done with 10, 8, 6, 4, 2 and 0 ppm solution of the metals obtained by serial dilution of 1000 ppm of the stock standard metal solution. Analysis of each sample was carried out in triplicate and results expressed in mg/kg (ppm). Data obtained was analysed and the results were expressed as mean ± Standard Deviation.

The heavy metals analysed are the most predominant metals in the region where the study took place (Appiah-Opong *et al.*, 2021).

# Determination of physico-chemical parameters of water

Water analyses were carried out using the standard procedures outlined in the standard methods for the examination of water (Rodier *et al.*, 2009). Physical parameters such as total dissolved solids (TDS) and turbidity were determined in addition to chemical parameters such as pH, Electrical Conductivity, BOD, DO and Salinity. Turbidity was measured using turbidimeter (Model HACH 2100P) and total dissolved solids (TDS) measured with a portable digital TDS meter (Model HI 99301) whilst salinity was measured using a hand-held refractometer. The physico-chemical parameters, including pH, Organic Carbon (OC), Conductivity and Cation Exchange Capacity (CEC) of soil were analysed using HACH model multi-probe meter (U – 50 series).

# Public health risk assessment of heavy metals in fish

Researchers have used different health risk estimation methods to evaluate the risk that consumption of contaminated fish poses to humans (Kwaansa-Ansah et al., 2019; Rai et al., 2019; Griboff et al., 2017 and Yi et al., 2017). In order to identify the number of pollutants consumed on a daily basis, the Estimated Daily Intake (EDI) was used (Kwaansa-Ansah et al., 2019 and Vrhovnik et al., 2013). Similarly, Vrhovnik et al. (2013) reported that the EDI of potentially toxic elements (PTE) is directly proportional to PTE concentrations in the food consumed daily.

### Data analysis

The means, maximum and minimum ranges for various parameters were generated from data using Statistical Package for the Social Sciences (SPSS) software version 22.0. Analysis of variance (ANOVA) at 95% confidence level (5% level of significance) was used to test for the significant differences in concentrations among the various water sources as well as fish species. The Estimated Daily Intake (EDI), hazard quotient (HQ) and Target Hazard Quotients (THQ) were determined for the fish. Karl Pearson's Product Moment Correlation analysis was carried out to establish the strength and direction of the relationship between the parameters of the water samples and soil samples. A correlation coefficient of p < 0.5 was deemed statistically significant.

The EDI of heavy metals for adults was calculated as follows (Rai et al., 2019; Kwaansa-Ansah et al., 2019; and Vrhovnik et al., 2013):

$$EDI = \frac{C \times C cons}{Bw}$$
 Eq. 1

where C is the concentration of heavy metals in fish (mg/kg wet weight), C  $_{\rm cons}$  is the national average daily

consumption of fish (71 g/day Bw), and Bw represents the body weight (Adult = 70 kg; Chn (1-11 yrs) = 30kg).

$$HQ = \frac{EDI}{RfD}$$
 Eq. 2

$$THQ = \frac{\text{EF x ED x FIR x C}}{\text{RfD x W x ATn}} \times 10^{-3}$$
 Eq. 3

Where:

EF = the exposure frequency (365 days/year)

ED = the exposure duration (64 years (WHO, 2020)) equivalent to the average lifetime in

Ghana;

FIR = the food ingestion rate (in grams per person per day); Ghana's average annual per capita consumption of fish is at 26 kg (and the daily ingestion works out to 71 g/day) (USDA, 2019).

C = the metal concentration in food (mg/g);

RfD = the oral reference dose (in milligram per kilogram per day);

W = the average body weight (71 kg for adults (Shirazu et al., 2017), and

ATn = the average exposure time for non-carcinogens (365 days/year i.e., number of exposure years, assuming 64 years in this study). It was assumed that cooking does not affect the toxicity of heavy metals in seafood.

### **RESULTS AND DISCUSSION**

### Physico-chemical parameters of water

A summary of the physico-chemical parameters of water is presented in Table 1. The mean pH values of water (main test seawater, hand-dug well and borehole water) sampled from the study area were generally higher than their controls. The pH values ranged from a minimum of  $6.1\pm0.35$  (borehole main test) to a maximum of  $7.6\pm0.24$  (sea main test). Analysis of variance (ANOVA) showed a significant difference (p < 0.05) in pH among the different water sources. Bonferroni pairwise multiple comparisons showed no significant difference (p > 0.05) between the main water sources and their controls. The results of pH values in all sources of water in this study were higher than those reported by Chiamsathit *et al.* (2020) in Thailand.

The mean conductivity of the water samples ranged from a minimum of  $246.8\pm88.9\mu$ S/cm (borehole) to a maximum of  $4618.9\pm673.0\mu$ S/cm (Seawater). Analysis of variance (ANOVA) showed a significant difference (p < 0.05) among water sources. Bonferroni pair-wise multiple comparisons showed no significant difference (p > 0.05) between the main water sources (borehole, handdug well and sea water) and their controls. The recorded conductivity value for water from hand-dug wells (main source) and the control samples exceeded the WHO recommended limit of 250  $\mu$ S/cm for drinking water. The shallow nature of the wells might have allowed

pollutants from the environment to enter the water. Oil and gas installations come with discharges such as produced water, process water, sewerage, sanitary and domestic waste, spills and leakages (EP Forum/ UNEP, 1997). The values reported in this study were lower than the values reported by Barmon *et al.* (2018) in water from Mokesh beel in Bangladesh except for values for sea water.

The BOD of water was generally higher in the main water sources than their controls. This suggests a relatively higher organic pollution of the main water source. Analysis of variance (ANOVA) of BOD revealed statistically significant differences (p < 0.05) among the different water sources. There was, however, no significant difference (p > 0.05) between the main test water sources and their controls.

The mean BOD values in the main water sources ranged from 4.14±0.48 mg/L (hand dug well) to 6.08±0.35mg/L (borehole main source).

Generally, DO in the main water test was higher than the controls. The mean DO of the main (test) water samples ranged from a minimum of 8.29±0.72mg/L (Hand-dug well) to a maximum of 10.37±0.31mg/L (borehole). Analysis of variance (ANOVA) showed no significant difference (p > 0.05) among the different water sources. There was no significant difference (p > 0.05) between the test water and their controls. Generally, a higher BOD in a sample translates into a lower DO. In Fishes, an average of 5 mg/L of dissolved oxygen (DO) is required for them to survive (Mallya, 2007). Thus, decreased DO levels may result in the death of fish and or stress of fishes and other organisms. Factors that affect DO levels in natural water include the presence of oxidisable substances, salinity, turbulence, temperature, and pressure (Demeke and Tassew 2016). DO values in this study were higher than the values (less than 6.0mg/l) reported by Barmon et al. (2018) in water from Mokesh beel in Bangladesh.

The results of water salinity recorded in the main test water were generally higher than their controls. However, there was no significant difference (p > 0.05) between the main water sources and their controls. The values of salinity in the main test water sources ranged from a minimum of  $0.008\pm0.04\,\mathrm{mg/L}$  (borehole) to a maximum of  $29.9\pm5.02\,\mathrm{mg/L}$  (sea water). Analysis of variance showed statistically significant differences (p < 0.05) in salinity among the different water sources.

Similarly, the mean values recorded for TDS in the test water ranged from a minimum of  $161.0\pm57.58$  mg/L (borehole) to a maximum of  $2766.7\pm400.67$ mg/L (sea water). Analysis of variance showed a significant difference (p < 0.05) among the different water sources but no significant difference (p > 0.05) between the test water and their controls. Values near 600 mg/L are acceptable, whereas those greater than 1000 mg/L are unpalatable for drinking (Apau *et al.*, 2014). The TDS

for all sources of water were within WHO recommended guidelines (1000mg/L and 30,000mg/L) for drinking water and saline water, respectively. The results of TDS from all water sources in this study are similar to those reported by Amfo-Out *et al.* (2012) and Chiamsathit *et al.* (2020) at the Ga East Municipality of Ghana and Thailand, except for the sea water.

The results of turbidity of the test water samples ranged from a minimum of  $0.64\pm0.11$ NTU (borehole) to a maximum of  $12.1\pm3.15$  NTU (sea water). The turbidity of test water samples was higher than their controls. Analysis of variance showed statistically significant differences (p < 0.05) in turbidity among the different water sources but no significant difference (p < 0.05) was recorded between the test water and their control.

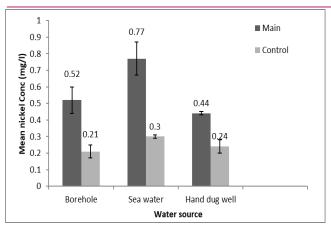
### Heavy metal concentration in water samples

The mean concentration of Nickel in test water ranged from a minimum of 0.44 mg/L (hand-dug well) to a maximum of 0.77mg/L (sea water) (Fig. 2). The value of Nickel in the main water sources was higher than their controls. Analysis of variance, however, showed no significant difference (p > 0.05) between the main test water and their respective controls.

The presence of Cr was recorded in the different sampled water. The mean Cr levels of the test water samples ranged from a minimum of 0.07 mg/L (sea water) to a maximum of 0.12 mg/L (borehole) (Fig. 3). The levels of Cr in the test water samples showed higher variations compared to their controls. However, analysis of variance showed no significant differences (p > 0.05) between the main test water and their respective controls.

Similarly, sea water recorded the highest value (64.54 mg/L) of Fe with the least (0.31 mg/L) recorded in the borehole (Fig. 4). There were however no significant differences in Fe among the water sources.

The higher levels of heavy metals recorded in the main test water may be attributed to anthropogenic activities as heavy metals may be associated with oil and gas activities. Also, Iron with a permissible level of 10mg/l had a mean value far above that of the control sample. Organic substances which form a major part of crude oil and petroleum products may to a large extent increase the heavy metal concentration upon release into water bodies. Meanwhile, smaller amounts of heavy metals such as lead (Pb), Cadmium (Cd), Mercury (Hg), Chromium (Cr), Cobalt (Co), Iron (Fe) and Copper (Co) above recommended levels are known to affect humans and the aquatic ecosystem (Ademoroti, 1996). According to Adesiyan et al. (2018), traces of heavy metals such as Cr, Pb, Fe, and Hg above stipulated levels are toxic to aquatic ecosystems and human well-being. Amoasah (2010) also revealed that produced water for ocean discharge contains up to 48 parts per million (ppm) of petroleum because it



**Fig. 2.** Mean Nickel levels of water samples from Western Coast of Ghana.

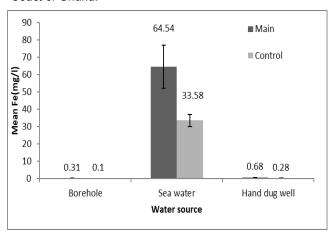


Fig. 4. Mean Iron levels of water samples from Western Coast of Ghana.

usually has contact with crude oil in the reservoir rocks. All water samples were within the permissible limit of guideline values of (WHO, 2011) except for sea water.

### Heavy metals concentration in fish samples

All heavy metals (Pb, Ni and Fe) analyzed in the two species (*Pomatomus saltatrix and Cynoscion regalis*) were present at varying concentrations except for Cd and Cr that were not detected. Generally, heavy metal concentration in *Cynoscion regalis* was higher than in *Pomatomus saltatrix* for all metals detected. The mean concentration (mg/kg) of Pb, Ni and Fe in *Pomatomus saltatrix fish* were  $1.86 \pm 0.19$ ,  $1.95 \pm 0.25$  and  $132.82 \pm 4.31$ (mg/kg) respectively while those in *Cynoscion regalis* species were  $2.44 \pm 0.13$ ,  $3.13 \pm 0.12$  and  $200.26 \pm 5.77$  respectively (Table 2). Student t-test statistics showed significant differences (p < 0.05) in concentrations of heavy metals between the two fish species (Table 3).

The higher mean values of heavy metals in the fish could indicate prevalence from the sampling sites. The concentration of Pb in fish samples in this study is higher than the one reported by Maurya, *et al.* 

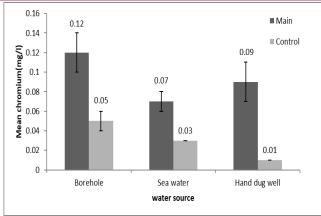


Fig. 3. Mean Chromium levels of water samples from Western Coast of Ghana.

(2019) in the River Ganga basin but lower than the concentration reported by Kortei et al. (2020) from Ankobrah and Pra basins in Ghana. According to Burger et al. (2002), there are two main routes of heavy metals exposure in fish species. The primary route of intake of these metals is through the gills or transport of dissolved contaminants in water across biological membranes and ionic exchange. The secondary route is through ingestion of food or sediment particles with subsequent transport across the gut. The aquatic micro flora or micro fauna which constitutes the food chain for fish species accumulate these metals in their living cells from the environment and as fish consume them they bio-accumulate these heavy metals which may subsequently affect humans in the food chain. These metals are known to produce adverse effects on aquatic biota and human health (Akan et al., 2012). Yang et al. (2010) reported that fish is very sensitive to Pb and its uptake increases with increasing concentration in the environment. Abarshi et al. (2017) reported that a fish living in contaminated sediment accumulates higher concentrations of Pb in gills and muscles, which is in line with this finding. The increase in contamination of metals in fish may be due to metalcontaminated diet, which may come from the discharge of oil residue and produced water that contains high levels of heavy metals (Owamah, 2013). The concentration of the heavy metals in the fish species increased in the order of Pb < Ni < Fe.

### Health risk assessment

Fish consumption has many health benefits; however, consuming toxic contaminated fish can have detrimental health effects on humans.

The calculated EDI, HQ, THQ for adults and children are presented in Table 5. For EDI the value for adults ranged from 1.88 x 10<sup>-3</sup> to 2.03 x 10<sup>-1</sup> (mg/kg bw/day) while that for children ranged from 4.40 x 10<sup>-3</sup> to 4.74 x 10<sup>-1</sup> (mg/kg bw/day). Similarly, the hazard Quotient val-

Table 1. Physico-chemical parameters of water sampled from the West coast of Ghana.

Water source	рН	<b>EC</b> (μS/cm)	BOD (mg/L)	<b>DO</b> (mg/L)	Salinity (ppt)	TDS (mg/L)	Turbidity (NTU)
Borehole main test	6.1±0.35 <sup>a</sup>	246.8±88.9 <sup>a</sup>	6.08±0.35 <sup>a</sup>	10.37±0.31 <sup>a</sup>	0.008±0.04 <sup>a</sup>	161.0±57.58 <sup>a</sup>	0.64±0.11 <sup>a</sup>
Sea water main test	7.6±0.24 <sup>b</sup>	4618.9±673.0 <sup>b</sup>	4.31±0.20 <sup>b</sup>	8.42±0.76 <sup>a</sup>	29.9±5.02 <sup>b</sup>	2766.7±400.67 <sup>b</sup>	12.1±3.15 <sup>a</sup>
Hand dug well main test	6.8±0.19 <sup>a</sup>	623±107.5 <sup>a</sup>	4.14±0.48 <sup>a</sup>	8.29±0.72 <sup>a</sup>	0.30±0.06 <sup>c</sup>	399.7±67.53°	0.87±0.35 <sup>a</sup>
Control Borehole	7.3±0.15 <sup>a</sup>	133.3±53. <sup>6a</sup>	4.55±0.78 <sup>a</sup>	13.4±1.88 <sup>a</sup>	0.01±0.00 <sup>a</sup>	67.7±26.43 <sup>d</sup>	0.10±0.09 <sup>a</sup>
Control Sea water	8.0±0.20 <sup>b</sup>	2900±520.0°	1.80±0.43°	10.00±0.21 <sup>a</sup>	29.2±6.80 <sup>b</sup>	1440.0±160.00 <sup>e</sup>	10.20±7.78 <sup>a</sup>
Control Hand dug well	7.6±0.21 <sup>b</sup>	350±70.0°	2.73±0.35 <sup>b</sup>	11.6±2.37 <sup>a</sup>	0.24±0.01 <sup>a</sup>	170.0±30.00 <sup>a</sup>	0.22±0.04 <sup>a</sup>

Data on the same row with different superscript (a, b, c,) are significantly different at p < 0.05

Table 2. Concentrations of heavy metals in P. saltatrix and C. regalis species from the study area.

Heavy metals (mg/kg)	Fish species	Mean ± SD	Min.	Max.
Lead	Pomatomus saltatrix	1.86 ± 0.19	1.54	2.22
Leau	Cynoscion regalis	2.44 ± 0.13	2.18	2.62
Cadmium	Pomatomus saltatrix	ND	0.00	0.00
	Cynoscion regalis	ND	0.00	0.00
Chromium	Pomatomus saltatrix	ND	0.00	0.00
	Cynoscion regalis	ND	0.00	0.00
Nickel	Pomatomus saltatrix	1.95 ± 0.25	1.45	2.24
Mickel	Cynoscion regalis	3.13 ± 0.12	2.89	3.28
Iron	Pomatomus saltatrix	132.82 ± 4.31	124.50	138.96
	Cynoscion regalis	200.26 ± 5.77	190.00	210.00

ND = Not Detected

**Table 3.** Student t – test analysis of heavy metals in *P. saltatrix and C. regalis* fish samples from the Western Coast of Ghana.

	t-value	df	p-value	e Mean	95 % Confidence level		
	t value	<b>W</b> 1	p value	difference	Lower	Upper	
Lead	6.854	5	0.001*	1.15000	0.72	1.58	
Nickel	5.260	5	0.003*	1.54333	0.79	2.29	
Iron	10.734	5	0.000*	165.5433	125.89	205.19	

<sup>\*</sup>Significant (P≤0.05).

ue for adults ranged from 0.099 to 0.537, while that for children ranged from 0.231 to 1.649. The Target Hazard Quotient values for adults ranged from 0.036 to 0.258 while values for children ranged from 0.084 to 0.864. The calculated value for Total Targeted Hazard Quotient (TTHQ) for adults who consume these fish species were less than 1 (< 1) (Table 3) however, those for children were greater than 1 (>1) for Cynoscion regalis fish species but less than 1 (< 1) for Pomatomus saltatrix species (Table 4). The TTHQ values (2.307) for children increases even much higher if they consume both fish species. Exposure to high concentrations of Pb, Ni and Cr are reported to cause central nervous system disorders, cardiovascular diseases, kidney failure, lung fibrosis, fertility problems, sinus node dysfunction, and atrioventricular conduction disturbances (Renieri et al., 2019; WHO, 2019 and Wani et

al., 2015). Similarly, high concentrations of Hg in human is reported to cause kidney disease, liver failure, and memory loss (WHO, 2017). Similarly, Ho-Wang & Wenxia (2020) reported that Fe toxicity causes Liver necrosis, Myocardial dysfunction and Cardiogenic shock among others.

### Conclusion

This study assessed heavy metal concentrations in water and two fish species from the western coast of Ghana. Results showed appreciable concentrations of Pb, Ni, and Cr in the water samples. A high amount of Pb was detected in the muscles of P. saltatrix and C. regalis. The concentration of heavy metals in the fish species increased in the order Pb < Ni < Fe. The EDI value for adult humans ranged from  $1.88 \times 10^{-3}$  to 2.03

**Table 4.** Hazard Quotients and Target Hazard Quotients for Heavy Metals of *P. saltatrix* and *C. regalis* species from the Western Coast of Ghana.

Metals	Fish species	EDI(adults) (mg/kg bw/day)	EDI(chn) (mg/kg bw/day)	HQ (adults)	HQ (Chn)
Lood	Pomatomus saltatrix	1.88 x 10 <sup>-3</sup>	4.40 x 10 <sup>-3</sup>	0.537	1.257
Lead	Cynoscion regalis	2.47 x 10 <sup>-3</sup>	5.77 x 10 <sup>-3</sup>	0.706	1.649
Cadmium	Pomatomus saltatrix	-	-	-	-
	Cynoscion regalis	-	-	-	-
Chromium	Pomatomus saltatrix	-	-	-	-
	Cynoscion regalis	-	-	-	-
Nickel	Pomatomus saltatrix	1.98 x 10 <sup>-3</sup>	4.62 x 10 <sup>-3</sup>	0.099	0.231
Nickei	Cynoscion regalis	3.17 x 10 <sup>-3</sup>	7.41 x 10 <sup>-3</sup>	0.158	0.371
Iron	Pomatomus saltatrix	1.34 x 10 <sup>-1</sup>	3.12 x 10 <sup>-1</sup>	0.191	0.445
	Cynoscion regalis	2.03 x 10 <sup>-1</sup>	4.74 x 10 <sup>-1</sup>	0.290	0.677
Total		3.47 x 10 <sup>-1</sup>	8.1 x 10 <sup>-1</sup>	1.981	4.630

 $\Sigma HQs = HI, HI_{adults} = 1.981; HI_{chn} = 4.630$ 

**Table 5.** Estimated Total Target Hazard Quotient (TTHQ) of *P. saltatrix* and *C. regalis* fish species from the Western Coast of Ghana.

Fish species	Population	TTHQ
Pomatomus saltatrix	Adult	0.303
Pomatomus saltatrix	Children	0.706
Cynoscion regalis	Adult	0.422
Cynoscion regalis	Children	1.601

Total target hazard quotients for both fish species for adults (TTHQ $_{\text{adults}}$ ) = 0.725

Total target hazard quotients for both fish species for children (TTHQ $_{\text{chn}}$ ) = 2.307

x  $10^{-1}$  (mg/kg bw/day) while that for children ranged from 4.40 x  $10^{-3}$  to 4.74 x  $10^{-1}$ (mg/kg bw/day). Health risk assessment showed that the TTHQ for adult was less than 1 (< 1) however those for children were greater than 1 (>1) for *C. regalis* species but less than 1 (< 1) for *P. saltatrix* species which means that children who eat these fishes are exposed to health risk in their lifetime. The general public's awareness and education on the threshold of heavy metals in fish and water are crucial to safeguard human health.

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### **Conflict of interest**

The authors declare that they have no conflict of interest.

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